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Highly Diastereoselective [3+2] Cyclopenta[b]annulation of Indoles with 2-Arylcyclopropyl Ketones and Diesters

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Dedicated to Prof. Dr. Hartmut Laatsch on the occasion of his 60th birthday

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A highly diastereoselective Lewis acid (BF $_3$ -Et $_2$ O or TiCl $_4$) induced [3+2] cycloaddition of substituted and unsubstituted indoles with 2-arylcyclopropyl ketones/diesters yielding cyclopenta[b]indoles in high yields is reported. This methodology has also been extended to tetrahydrocarbazole, cy-

clopenta[b]- and cyclohepta[b]indoles affording tetracyclic propellane type frameworks in modest yields.

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Introduction

Cyclopenta[b]indole ring systems occur in a number of indole alkaloids, [1] notably tremorgenic mycotoxins such as paxilline, lolitrems, penitrems, janthitrems, and paspaline^[2] and in monoterpenoid alkaloid vuehchukene^[3] which has been shown to exhibit mixed estrogen and antiestrogen, as well as potent antiimplantation, activities.^[4–6] Recently a cyclopenta[b]indole derivative has been identified as a promising prostaglandin D₂ (PGD₂) receptor antagonist in the alleviation of various allergic disorders.^[7] Another class of naturally occurring cyclopenta[b]indole alkaloids are represented by kopsane, [8] possessing an architecturally beautiful tetracyclic propellane type annulated indole substructure exhibiting cholinergic activity. [9,10] Several syntheses of yuechukene^[11–13] and cyclopenta[b]indole frameworks have been reported in the literature.[14-19] Kerr and co-workers recently described an elegant approach for cyclopenta[b]indoles by reaction of 1,3-dimethylindoles with 1,1-cyclopropane diesters in the presence of ytterbium trifluoromethanesulfonate.[20-21] They further elaborated this reaction for the synthesis of substrates possessing a tetracyclic core present in kopsane and related alkaloids by reacting tetrahydrocarbazole with cyclopropane diesters. The presence of a 3-methyl substituent in the indole ring was necessary for the success of this cyclopenta[b]annulation to avoid deprotonation and rearomatization of the putative imminium ion intermediate formed by attack at the 3-position of indole on cyclopropane diesters. With 3-unsubstituted indoles, the product formed was one in which alkylation took place to yield 2-alkoxycarbonyl-4-(3-indolyl)butanoate.[22,23] Also with N-unsubstituted indoles, these reactions were complicated by the formation of N-alkylated products by attack of the nitrogen atom of indole on a second equivalent of the cyclopropane diester, thus requiring N-protection of indole. During the course of our continued work on carbocationic rearrangements of a range of cyclopropyl ketones, [24] we examined the formal [3+2] cycloaddition of the indole 2,3double bond with few aryl cyclopropyl ketones and observed that 1,3-unsubstituted indoles react with 4-methoxyphenyl-substituted cyclopropyl ketones in the presence of BF₃·Et₂O to give substituted cyclopenta[b]indoles in high yields and with complete diastereoselectivity. We herein describe the results of these studies in this paper.

Results and Discussion

The cyclopropyl ketone **2a** derived from (4-methoxybenzylidene)acetophenone was selected for this model study because of its donor-acceptor character due to the presence of the cation stabilizing 4-methoxyphenyl group. Initially, a study was undertaken to probe the optimal conditions for the annulation reaction (Scheme 1). Table 1 illustrates the results of our studies in which several Lewis acids [BF₃·Et₂O, Yb(OTf)₃, SnCl₄, and TiCl₄] were surveyed under varying conditions. Best results were obtained with BF₃·Et₂O in nitromethane (entry 5) which gave the cyclopentannulated indole **3a** in 92% yield within 2 h, surprisingly as a single diastereomer, in contrast with Kerr's obser-

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vation. The structure and stereochemistry of the cyclopentannulated indole 3a was established with the help of spectral and X-ray crystallographic data (Figure 1). Encouraged by these results, we undertook a detailed study of

Scheme 1.

Table 1. Formation of cyclopenta[b]indole 3a under the influence of various Lewis acids.

Entry	Lewis acid	Solvent	Conditions	Time [h]	Yield [%]
1	Yb(OTf) ₃ [a]	CH ₃ CN	0 °C to room temp.	36	31(1:1) ^[b]
2	$Yb(OTf)_3^{[a]}$	CH_2Cl_2	0 °C to room temp.	36	_[d]
3	BF ₃ ·Et ₂ O	CH_2Cl_2	0 °C to room temp.	5	42 (2:1) ^[b]
4	BF ₃ ·Et ₂ O	CH ₃ CN	0 °C to room temp.	3	48 (1:1) ^[b]
5	BF ₃ ·Et ₂ O	CH_3NO_2	0 °C to room temp.	2	92 ^[c]
6	SnCl ₄	CH_2Cl_2	0 °C to room temp.	6	28 (2:1) ^[b]
7	SnCl ₄	CH_3NO_2	0 °C to room temp.	6	31 (4:1) ^[b]
8	TiCl ₄	CH_2Cl_2	−78 °C to room temp.	4	70 ^[c]
9	TiCl ₄	CH_2Cl_2	0 °C to room temp.	5	42 (5:1) ^[b]

[a] 10 mol-% catalyst. [b] Mixture of diastereomers. [c] Single diastereomer. [d] No reaction.

this reaction by varying the substituents on the indole ring as well as on the aryl group of cyclopropyl ketones with a view to examine the scope of this [3+2] cycloaddition reaction. These results are summarized in the Scheme 2 and Table 2. Thus, 5-bromoindole 1b reacted smoothly with cyclopropyl ketone 2a furnishing the cyclopenta[b]indole 3b in 70% yield as a single diastereomer (entry 2). The cyclization was equally facile with N-methylindole and the product 3c was obtained in 83% yield (entry 3). The presence of a methyl group either at the 2- or 3-position of the indole (1d and 1e) did not affect the course of reaction and the cycloadducts 3d-e were obtained in 80% and 83% yields, respectively, as single diastereomers (entries 4–5). The Xray crystallographic studies of 3d also displayed all three tertiary and quaternary hydrogens syn to the 2-methyl group. Similarly, the corresponding 2,3-dimethylindole 1f also underwent facile reaction with 2a under identical conditions yielding the sterically crowded annulated indole 3f in 93% yield, having two vicinal quaternary centers adjacent to two tertiary centers (entry 6). On the other hand, the corresponding 1,2,3-trimethylindole 1g failed to react with the cyclopropyl ketone 2a under varying conditions, yielding only the indole 1g along with a polymeric product (entry 7).

Surprisingly the cyclopropyl ketone 2b from (3,4-dimethoxybenzylidene)acetophenone failed to yield the cyclopentannulated product when reacted with indole 1a in the presence of $BF_3 \cdot Et_2O$ in nitromethane. The product isolated from the reaction mixture was characterized as the

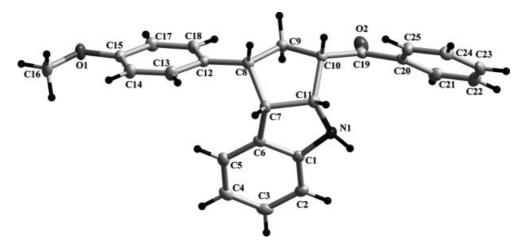


Figure 1. Molecular structure of 3a shown at 50% ellipsoidal probability level.

$$X = H, Br$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R^{6}$$

$$R^{7}$$

Scheme 2

Table 2. Cyclopentannulation of indoles 1 with cyclopropyl ketones 2.

Entr	у 1	2	Product	% Yield [[]
	1		X H H NC6H4-4-MeO	
	6I		Ph	
		Y	Н Н О	
[b]	1a, X		3a, X = H	92
2 ^[b]	1b, X	- Br 2a	3b, X = Br H	70
	F		H NC6H4-4-MeO	
		1	Ph	
	1	/le	N H H O	
3 ^[b]	1c	2a	3c	83
	1	7	H NC ₆ H ₄ -4-MeO	
	The same	Me	Ph	
	H	ì	N Me H O	
4 ^[b]	1d	2a Me	3 d H	80
	5	1	Me NC6H4-4-MeO	
	The same of the sa	y		
5 ^[b]	i i		й н н о	- 0.0
2101	le	2a	3e HC _e H₄-4-MeO	83
	F	-√ ^{Me}	Ph	
	The same	N Me	N Me H O	
6 ^[b]	lf l	2a	H Men O	93
	F	Me		
	W.	Me	-	_[h]_
		de		
7 ^[b]	1g	2a	n-E	Bu
	F	\mathbb{R}^3	R ³ H	2
	The same of the sa	R^2	Ph	ie
	N I R	1	N R ² H O	
g[c]	1a	2b	$ \begin{array}{cccc} N & R^2 & H & O \\ R^1 & & & & \\ \mathbf{5a}, R^1 = R^2 = R^3 = H \end{array} $	84
9[c]	1e	2b	5e, $R^1 = Me$; $R^2 = R^3 =$	
0[c]	1f	2b	5f , $R^1 = H$; $R^2 = R^3 = N$	
			Me. H. Ph Me. Ph	
		[Ph + Ph	Ph
		"	H H H H	
[][b]	1e		н о	85 ^[a]
	ie	2e	6e" (2:1)	831-1
			Me H, Ph Me Ph	
			Me Me	Me
			H H O N H H	100
2 ^[b]	1e	2d	7e' 7e" (3:1)	70
			Ph Ph	
			H, R H	
			H Ph	
			H H H O	
3[6]	1a		8 9a, R ¹ = H	88
[4 ^[b]	1e		8 9e, R ¹ = Me	79

[a] Yield of pure isolated product. [b] Reaction conditions: BF₃·Et₂O, 0 °C to room temp., 2 h. [c] Reaction conditions: TiCl₄, CH₂Cl₂, –78 °C, 2 h. [d] Indole **1g** was recovered fully. [e] Pure **6e**′ was separated in 45 % yield.

1,4-diaryl-3,4-dihydrocarbazole **4b** (75%) on the basis of spectral and analytical data (Scheme 3). However, on

changing the Lewis acid to TiCl₄ at lower temperature (-78 °C), the reaction proceeded smoothly in dichloromethane yielding the cyclopentannulated adduct 5a in 84% yield as single diastereomer (Scheme 2, Table 2, entry 8). Similarly, the corresponding 1-methyl- (1c) and 2,3-dimethylindole (1f) also underwent smooth cycloaddition with the 2-(3,4-dimethoxyphenyl)cyclopropyl ketone 2b in the presence of TiCl₄/CH₂Cl₂ catalyst to afford cyclopentaannulated indoles 5c and 5e in 85% and 83% yields, respectively, in a highly stereocontrolled fashion (Table 2, entries 9–10). Cyclopentannulation of the indole 1a with phenyl 2-phenylcyclopropyl ketone (2c), with lesser cationic stabilization in comparison with 2a, was investigated next. Thus, the reaction of 1a with 2c in the presence of either BF3. Et2O or TiCl₄ or other Lewis acids [Yb(OTf)₃, SnCl₄] failed to yield the corresponding cyclopentannulated indole adduct and the only product isolated (with either BF₃·Et₂O or TiCl₄) was found to be the 1,4-diaryl-3,4-dihydrocarbazole 4c (Scheme 3), whereas with Yb(OTf)₃/CH₃CN the indole 1a was recovered unchanged even after prolonged reaction time (48 h). On the other hand, the corresponding 3-methvlindole 1e underwent smooth cyclopentaannulation with ketone 2c to give the cycloadduct 6e in 85% overall yield as mixture of two diastereomers 6e' and 6e'' (2:1) (entry 11, Table 2). Similarly, the corresponding 1-phenyl-2-acetylcyclopropane 2d also reacted with 3-methylindole 1e under identical conditions to give the diastereomeric mixture of cycloadducts 7e' and 7e'' (3:1) in 70% overall yield (entry 12, Table 2).

^a BF₃·Et₂O, CH₃NO₂, 0 °C to r.t., 89%

^b TiCl₄, -78 °C to r.t., CH₂Cl₂

Scheme 3.

Cyclopentaannulation of the indole 1a with 1-(styrenyl)-2-benzoylcyclopropane 8 was examined next with a view to probe the effect of the styrenyl group towards cationic stabilization during the ring opening of the cyclopropyl ketone 8 in the presence of Lewis acid. Thus, when the indole 1a and phenyl 2-(α -styryl)cyclopropyl ketone (8) were exposed to $BF_3 \cdot Et_2O$ in nitromethane under standard reaction conditions, the styrenyl-substituted cycloadduct 9a was formed in 88% yield as a single diastereomer with the four cyclopentane hydrogens having syn stereochemistry (Table 2, entry 13). Similarly, the corresponding 3-methylindole 1e also yielded the styrenyl-substituted cyclopenta-[b]indole 9e in 79% yield with high stereoselection (Table 2, entry 14).

Next we extended our studies to the cycloaddition of indoles with aryl and styrenyl-substituted cyclopropyl diesters 10a-b and 13 (Scheme 4, Table 3). The presence of two electron-withdrawing substitutents at the acceptor carbon is expected to facilitate the cationic ring opening of arylcyclopropanes in the presence of Lewis acids. Thus, when the indole 1a was treated with 4-methoxyphenylcyclopropyl diester 10a, in the presence of BF₃·Et₂O, analysis of the product mixture revealed formation of both the diastereomers (11a' and 11a'') of the cyclopenta[b]indole 11a (2:1) in overall yield of 86% (Scheme 4, Table 3, entry 1). Similarly, the N-methylindole 1c also afforded the adduct 11c (90%) in 1:1 diastereomeric ratio when reacted with cyclopropyl diesters 10a under identical conditions (Scheme 4, Table 3, entry 2). On the other hand, cyclopentannulation of 3methylindole 1e with either 4-methoxyphenyl or phenylsubstituted cyclopropyl diesters (10a or 10b) proceeded in highly diastereoselective fashion yielding only diastereomers 11e and 12e in 89% and 72% yields, respectively (Table 3, entries 3 and 4). The stereochemical assignment of the adducts 11e and 12e, having all three contiguous hydrogens syn to each other, was established with the help of NOESY correlation spectra. The reaction of styrenyl-substituted cyclopropyl diester 13 with the indole 1a in the presence of either BF₃·Et₂O or TiCl₄ gave only a complex mixture of products, whereas use of Yb(OTf)₃ in CH₃CN furnished the cycloadduct 14 in 49% yield (Table 3, entry 5).

Scheme 4.

The success of the cyclopenta[b]annulation of the various substituted indoles, especially 2,3-dimethylindole 1f, prompted us to extend this reaction to tetrahydrocarbazole 15 as a dipolarophile with a view to synthesize propellane type annulation products (Scheme 5). Thus, under optimized reaction conditions, the reaction of tetrahydrocarbazole with cyclopropyl ketone 2a in the presence of BF₃·Et₂O in CH₂Cl₂ afforded a product (58%) which was characterized as the propellane 16. The product was formed as a single diastereomer (Scheme 5) and is in contrast to the results of Kerr with phenylcyclopropyl diester 12e yielding a 1:1 mixture of diastereomers. Attempts to isolate the other diastereomer from the reaction mixture were not successful.

The reaction of the corresponding cyclopenta[b]indole 17 with the 2-arylcyclopropyl ketones 2a and 2c also afforded the propellane derivatives 18 and 19 as single diastereomers in modest yields under similar conditions (Scheme 5). Similarly, the propellane synthesis was also extended to cy-

Table 3. Cyclopentannulation of indoles 1 with cyclopropyl diesters 10.

Entry 1	10	Product		% Yield ^[a]
		H HC ₆ H ₄ -4-MeO CO ₂ Et + CO ₂ Et	C ₆ H ₄ -4-M II J., H CO ₂ E	Et
1 ^[b] 1a	10a	11a'	11a" (2:1)	86
		HC ₆ H ₄ -4-MeO N H CO ₂ Et + Me	CO ₂ E	Εt
2 ^[b] 1c	10a	11c'	11c" (1:1)	90
		Me H CO ₂ Et CO ₂ Et		
3 ^[b] 1e	10a	$\mathbf{11e}, \mathbf{R}^3 = \mathbf{MeO}$		89
4 ^[b] 1e	10b	12e , $R^3 = H$	_	72
	H EtO ₂ C CO ₂ Et	Ph CO ₂ Et H CO ₂ Et		
5 ^[e] 1a	13	14	_	49

[a] Yields of pure isolated products. [b] Reaction conditions: $BF_3 \cdot Et_2O$, CH_3NO_2 , $0 \,^{\circ}C$ to room temp., $2 \, h$. [c] $10 \, mol - \% \, Yb(OTf)_3$, CH_3CN , room temp., $14 \, h$.

Scheme 5.

clohepta[b]indole 20 which on reaction with the cyclopropyl ketone 2a afforded two products of which the minor one (27%) was characterized as the propellane derivative 19 as a single diastereomer, whereas the structure of the major product (45%) was established as 22 on the basis of spectral and analytical data (Scheme 5).

In the ¹H NMR spectra of all newly prepared cyclopenta[2,3-b]indoles 3a-f, 5a, 5c, 5f, 6e, 7e, 9a, and 9e, the trans coupling constants (JH1H2", JH2"H3) and the geminal coupling constants (JH2'H2") are in the range of 12-13 Hz, whereas the three *cis* coupling constants $(JH^1H^2)'$, JH2'H3 and JH3H3a) are in the range of 5-6.1 Hz (Figure 2). The trans geminal proton (H2'') appears as a quartet with coupling constants between range of 12-13 Hz in all the ¹H NMR spectra of these compounds. On the other hand, the cis protons H2' and H3 appear as triplets of doublets with two middle peaks merging (showing five peaks) with two coupling constants in the range of 5-6.1 Hz (triplet) and 12-13 Hz (doublet). The cis coupling constants $JH^{1}H^{8b}$ and $JH^{3a}H^{8b}$ are in the range of 8.5–10.1 Hz. The H¹ proton appears as a difficult to analyze multiplet in all the ¹H NMR spectra. The same trend is also observed in the ¹H NMR spectra of products 11a, 11c, 11e, 12e, and 14 $(JH^{2''}H^3 \text{ and } JH^{2'}H^{2''} = 13.3-14.6 \text{ Hz})$, 16, 18, 19, and 21. Average values of coupling constants are reported.

Figure 2. Proton designation for cyclopenta[2,3-b]indole ring.

Mechanism and Stereochemistry

The probable mechanism for this unprecedented [3+2] annulation of 3-unsubtituted indoles (1a-d) with cyclopropyl ketone 2a is shown in the Scheme 6. Apparently the formation of stable zwitterionic intermediates via facile ring opening of cyclopropyl ketone is the key factor for this facile [3+2] cycloaddition of 3-unsubstituted indoles to cyclopentannulated products 3 via the imminium ion intermediate 24 in a kinetically controlled process. In the absence of the cation stabilizing 4-methoxy group, the ring opening of cyclopropyl ketone is slow and the resulting imminium ion 24 (Ar = Ph) undergoes slow tautomerization to indole 25 followed by its BF₃·Et₂O induced intramolecular cyclocondensation furnishing the 1,4-diaryl-3,4-dihydrocarbazole 4 as the exclusive product.

Scheme 6.

Another noteworthy feature of this [3+2] cycloannulation is the remarkable stereoselectivity which deserves comment. In all the cyclopentaannulation reactions of the indoles 1a– **g** and methoxyaryl-(or α -styryl-)cyclopropyl ketones **2a**-**b**, only one diastereomer was obtained which has a trans relationship between its angular substitutents (H or Me) and the substituents (Ar or ArCO) on the newly formed cyclopentane ring which are in an endo relationship with the bent ring system. The possible conformations for the transition states for the intramolecular cyclization of imminium ion intermediate 24 are shown in Figure 3. For the Z enolate the possible transition states are 24A and 24C whereas structures 24B and 24D represent possible transition states for the E enolate. Apparently the conformations 24C and **24D**, which will yield *trans* products, are of higher energy than 24A and 24B because of pseudodiaxial interactions shown in Figure 3. Therefore, all cis products 3 are formed through cyclization of intermediates 24A or 24B with minimal pseudodiaxial interactions.

Figure 3. Possible conformations for the formation of cyclopenta[2,3-*b*]indoles.

Conclusion

In summary, we have demonstrated an unusual [3+2] cycloannulation of 1,3-unsubstituted/substituted indoles with aryl-(or α -styryl-)cyclopropyl ketones under BF₃·Et₂O (or TiCl₄) induced reaction conditions yielding cyclopentannulated indolines in high yields and in a highly stereocontrolled fashion as single diastereomers. The reaction has also been extended to tetrahydrocarbazole, cyclopenta[b]- and cyclohepta[b]indoles affording tetracyclic propellane type annulated substructures present in kopsane type alkaloids. Further work to study the detailed mechanism and the scope of this annulation methodology are in progress.

Experimental Section

General: ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded in CDCl₃ and TMS was used as an internal refer-

ence. Melting points are uncorrected. Chromatographic purification was conducted by column chromatography using 100-200 mesh silica gel obtained from Acme Synthetic Chemicals. DMF and DMSO were distilled from CaH₂ and stored with molecular sieves. THF was distilled from sodium/benzophenone prior to use. NaH (as 60% suspension in oil), SnCl₄, BF₃·Et₂O (50% solution in Et₂O), TiCl₄, Yb(OTf)₃, and nitromethane (AR grade) were purchased from standard firms and used directly. The cyclopropyl ketones 2a, 2b, 2c, 2d, 8, and diesters 10a, 10b, and 13 were prepared according to the literature procedure^[25] in nearly quantitative yields. Single-crystal X-ray crystallographic data on 3a and 3d were collected at 100 K on a Bruker SMART APEX CCD diffractometer with graphite-monochromated Mo- K_{α} ($\lambda = 0.71073$ A) radiation using the ω - θ scan technique. The hydrogen atoms were fixed in their ideal positions and refined as a riding model except for the hydrogen atom of the chiral center, which was located from the difference Fourier map and refined isotropically. See Supporting Information for more details on X-ray crystal data and molecular structure of 3a and 3d.

General Procedure for the Cycloaddition of Cyclopropyl Ketones 2a, 2b, 2c, 2d, and 8 or Diesters 10a, 10b, and 13 to Indoles 1a-g, 15, 17, and 20 in the Presence of BF₃·Et₂O as a Lewis Acid: A solution of the Lewis acid (3.0 mmol) in dry solvent (nitromethane or CH₂Cl₂ or CH₃CN) (5 mL) was added dropwise at 0 °C to a stirred solution of the cyclopropyl ketone (2a, 2b, 2c, 2d, or 8 or diesters 10a, 10b, or 13) (2.0 mmol) and the appropriate indole (1a-g, 15, 17, or 20) (2.0 mmol) in respective solvent (10 mL). After the complete addition, the reaction mixture was stirred at room temp. for 1-4 h (monitored by TLC). It was then diluted with an ice-cold saturated NaHCO₃ solution (10 mL), and extracted by CHCl₃ $(2 \times 25 \text{ mL})$, washed with water $(1 \times 25 \text{ mL})$, followed by brine (25 mL) and dried with anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure to afford the cyclopentannulated indoles 3a-f, 6, 7, 9a, 9e, 11a, 11c, 11e, 12e, and 14, and dihydrocarbazoles 4b and 4c, or propellanes 16, 18, 19, and 21 which were purified by column chromatography using silica gel and hexane/ EtOAc as eluent.

 $(1R^*,3S^*,3aR^*,8bS^*)$ -[1-(4-Methoxyphenyl)-1,2,3,3a,4,8b-hexahydrocyclopenta[b]indol-3-yl]phenylmethanone (3a): Yield 92% (0.67 g); colorless crystals; m.p. 174–175 °C; R_f 0.66 (3:1 hexanes/ EtOAc). IR (KBr): $\tilde{v} = 3367$, 1676, 1602, 1512, 1463, 1362, 1258, 1178, 1034, 749 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 2.08$ (td, $J = 12.80, 5.8 \text{ Hz}, 1 \text{ H}, \text{H}^{2'}), 2.47 \text{ (q, } J = 12.80 \text{ Hz}, 1 \text{ H}, \text{H}^{2''}), 3.62$ (m, 1 H, H¹), 3.78 (s, 3 H, OCH₃), 3.93 (td, J = 12.80, 5.8 Hz, 1 H, H³), 4.12 (t, J = 9.6 Hz, 1 H, H^{8b}), 4.94 (dd, J = 9.6, 5.8 Hz, 1 H, H^{3a}), 5.79 (d, J = 7.3 Hz, 1 H, ArH), 6.28 (t, J = 7.3 Hz, 1 H, ArH), 6.42 (d, J = 7.8 Hz, 1 H, ArH), 6.77 (d, J = 8.5 Hz, 2 H, ArH), 6.85 (t, J = 7.5 Hz, 1 H, ArH), 6.96 (d, J = 8.5 Hz, 2 H, ArH), 7.51 (t, J = 7.0 Hz, 2 H, ArH), 7.59 (t, J = 7.3 Hz, 1 H, ArH), 8.04 (dd, J = 8.0 Hz, 1.2 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 30.9$, 48.8, 52.5, 54.4, 55.2, 65.2, 108.6, 113.2, 117.6, 126.6, 127.5, 127.6, 128.2, 128.8, 129.7, 131.5, 133.1, 137.2, 151.4, 158.2, 199.5 ppm. MS: m/z (%) = 369 (100) [M⁺]. C₂₅H₂₃NO₂ (369.46): calcd. C 81.27, H 6.27, N 3.79%; found C 81.11, H 6.09, N 3.54%.

(1*R**,3*S**,3a*R**,8b*S**)-[(4-Methoxyphenyl)-3a-methyl-1,2,3,3a,4,8b-hexahydrocyclopenta|*b*|indol-3-yl|phenylmethanone (3d): Yield 80% (0.61 g); colorless crystals; m.p. 170–171 °C; R_f 0.5 (6:1 hexanes/ EtOAc). IR (KBr): \tilde{v} = 3393, 1676, 1603, 1512, 1463, 1232, 1177, 1031, 828, 752, 707 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 1.50 (s, 3 H, CH₃), 2.17 (td, J = 12.5, 5.4 Hz, 1 H, H²′), 2.42 (q, J = 12.5 Hz, 1 H, H²′), 3.54 (d, J = 9.8 Hz, 1 H, H^{8b}), 3.67 (m, 1 H,

H¹), 3.74 (s, 3 H, OCH₃), 3.83 (dd, J = 12.5, 5.4 Hz, 1 H, H³), 4.86 (br. s, 1 H, NH), 5.89 (d, J = 7.3 Hz, 1 H, ArH), 6.26 (dt, J = 7.4, 1.0 Hz, 1 H, ArH), 6.51 (d, J = 7.5 Hz, 1 H, ArH), 6.76 (d, J = 8.6 Hz, 2 H, ArH), 6.83 (d, J = 8.6 Hz, 2 H, ArH), 6.90 (dt, J = 7.7, 1.0 Hz, 1 H, ArH), 7.47 (t, J = 7.2 Hz, 2 H, ArH), 7.56 (t, J = 7.3 Hz, 1 H, ArH), 7.96 (dd, J = 8.4, 1.1 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 27.7, 35.7, 49.1, 55.1, 57.0, 59.4, 72.3, 108.0, 113.0, 116.9, 126.7, 126.9, 127.5, 128.3, 128.5, 129.6, 131.7, 133.0, 138.2, 150.9, 158.0, 201.2 ppm. MS: m/z (%) = 383 (22) [M⁺], (100) [263]. C₂₆H₂₅NO₂ (383.48): calcd. C 81.43, H 6.57, N 3.65%; found C 81.31, H 6.34, N 3.42%.

(1R*,3S*,3aS*,8bS*)-[(4-Methoxyphenyl)-8b-methyl-1,2,3,3a,4,8bhexahydrocyclopenta[b]indol-3-yl]phenylmethanone (3e): Yield 83% (0.63 g); colorless crystals; m.p. 141–142 °C; $R_{\rm f}$ 0.60 (6:1 hexanes/ EtOAc). IR (KBr): $\tilde{v} = 3369$, 2957, 1674, 1600, 1512, 1461, 1358, 1301, 1248, 1185, 1029, 822, 744, 703 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 1.55 (s, 3 H, CH₃), 2.05 (td, J = 12.8, 5.6 Hz, 1 H, $H^{2'}$), 2.54 (q, J = 12.8 Hz, 1 H, $H^{2''}$), 3.19 (dd, J = 12.8, 5.6 Hz, 1 H, H¹), 3.77 (s, 3 H, OCH₃), 3.94 (td, J = 12.8, 5.6 Hz, 1 H, H³), 4.47 (d, J = 5.6 Hz, 1 H, H^{3a}), 5.68 (d, J = 7.3 Hz, 1 H, ArH), 6.29(t, J = 7.3 Hz, 1 H, ArH), 6.43 (d, J = 7.8 Hz, 1 H, ArH), 6.75 (d, J = 7.8 Hz, 1 H, ArH), 6.75 (d, J = 7.8 Hz, 1 H, ArH)J = 8.56 Hz, 2 H, ArH), 6.86 (t, J = 5.6 Hz, 1 H, ArH), 6.90 (d, J)= 8.5 Hz, 2 H, ArH), 7.51 (t, J = 7.4 Hz, 2 H, ArH), 7.59 (t, J = 7.4 Hz, 2 H, ArH)7.3 Hz, 1 H, ArH), 8.04 (d, J = 7.8 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 28.4, 32.5, 52.8, 55.1, 56.4, 57.8, 72.6, 108.3, 112.9, 117.4, 125.6, 127.3, 128.1, 128.8, 129.8, 131.3, 131.7, 133.0, 137.2, 150.8, 158.3, 199.6 ppm. MS: m/z (%) = 383 (100) [M⁺]. C₂₆H₂₅NO₂ (383.48): calcd. C 81.43, H 6.57, N 3.65%; found C 81.62, H 6.31, N 3.79%.

(1R*,3S*,3aS*,8bS*)-[(4-Methoxyphenyl)-3a,8b-dimethyl-1,2,3,3a,4,8b-hexahydrocyclopenta[b]indol-3-yl]phenylmethanone (3f): Yield 93% (0.73 g); yellow viscous oil; R_f 0.72 (19:1 hexanes/ EtOAc). IR (CH₂Cl₂): $\tilde{v} = 3397$, 2965, 2834, 2361, 1676, 1607, 1512, 1485, 1464, 1446, 1379, 1319, 1247, 1179, 1110, 1035, 829, 742 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 1.30$ (s, 3 H, CH₃), 1.34 (s, 3 H, CH₃), 2.07 (td, J = 12.5, 5.6 Hz, 1 H, H²), 2.35 (q, J= 12.5 Hz, 1 H, $H^{2''}$), 3.06 (dd, J = 12.5, 5.6 Hz, 1 H, H^{1}), 3.65 (s, 3 H, OCH₃), 3.70 (dd, J = 12.5, 5.6 Hz, 1 H, H³), 4.43 (br. s, 1 H, NH), 5.74 (d, J = 7.1 Hz, 1 H, ArH), 6.18 (t, J = 7.3 Hz, 1 H, ArH), 6.43 (d, J = 7.8 Hz, 1 H, ArH), 6.58 (d, J = 8.7 Hz, 2 H, ArH), 6.66 (d, J = 8.7 Hz, 2 H, ArH), 6.81 (t, J = 7.3 Hz, 1 H, ArH), 7.39 (t, J = 7.6 Hz, 2 H, ArH), 7.48 (t, J = 7.3 Hz, 1 H, ArH), 7.90 (dd, J = 8.5, 1.5 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 23.3, 35.6, 55.1, 55.6, 58.3, 58.9, 74.3, 107.6, 112.8, 116.8, 125.3, 127.5, 128.2, 128.5, 129.7, 130.0, 131.4, 131.9, 133.0, 138.1, 149.5, 158.2, 201.5 ppm. MS: m/z (%) = 397 (100) [M⁺]. C₂₇H₂₇NO₂ (397.20): calcd. C 81.58, H 6.85, N 3.52%; found C 81.70, H 6.72, N 3.62%.

(1*S**,3*S**,3*aR**,8*bS**)-Phenyl-(1-styryl-1,2,3,3a,4,8b-hexahydrocyclopenta|*b*|indol-3-yl)methanone (9a): Yield 88% (0.64 g); colorless solid; m.p. 150–151 °C; R_f 0.40 (9:1 hexanes/EtOAc). IR (KBr): $\tilde{v} = 3368$, 3027, 2926, 2374, 1679, 1599, 1482, 1364, 1322, 1251, 1150, 968, 745, 691 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 1.96$ (td, J = 12.5, 5.7 Hz, 1 H, H²'), 2.11 (q, J = 12.5 Hz, 1 H, H²'), 3.06–3.11 (m, 1 H, H¹), 3.82 (td, J = 12.5, 5.7 Hz, 1 H, H³a), 4.02 (t, J = 8.5 Hz, 1 H, H³b), 4.87 (dd, J = 8.5, 5.7 Hz, 1 H, H³aa), 5.96 (dd, J = 15.9, 9.0 Hz, 1 H, =CH), 6.42 (d, J = 15.9 Hz, 1 H, =CH), 6.44 (d, J = 7.5 Hz, 1 H, ArH), 6.55 (t, J = 7.3 Hz, 1 H, ArH), 6.89–6.94 (m, 2 H, ArH), 7.12–7.25 (m, 5 H, ArH), 7.42–7.54 (m, 3 H, ArH), 7.96 (d, J = 8.1 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 32.9$, 47.5, 51.8, 54.2, 65.6, 109.2, 118.0, 126.1, 126.5, 127.0, 127.5, 127.8, 128.2, 128.5, 128.8, 130.3, 131.0

133.1, 137.1, 137.3, 151.2, 199.4 ppm. MS: m/z (%) = 365 (100) [M $^+$]. C₂₆H₂₃NO (365.47): calcd. C 85.45, H 6.34, N 3.83%; found C 85.61, H 6.12, N 4.11%.

4-(3,4-Dimethoxyphenyl)-1-phenyl-4,9-dihydro-3*H***-carbazole (4b):** Yield 75% (0.57 g); colorless crystals; m.p. 177–178 °C; $R_{\rm f}$ 0.39 (6:1 hexanes/EtOAc). IR (KBr): $\tilde{v}=3338$, 1593, 1512, 1458, 1247, 1133, 1025, 759 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta=2.75$ (ddd, J=17.1, 10.2, 4.5 Hz, 1 H, H³), 2.95 (ddd, J=17.1, 8.2, 4.5 Hz, 1 H, H³), 3.77 (s, 3 H, OCH₃), 3.85 (s, 3 H, OCH₃), 4.37 (dd, J=10.2, 8.2 Hz, 1 H, H⁴), 5.97 (t, J=4.5 Hz, 1 H, =CH), 6.80 (d, J=8.3 Hz, 1 H, ArH), 6.88–6.95 (m, 4 H, ArH), 7.06 (dt, J=7.1, 2.4 Hz, 1 H, ArH), 7.19–7.26 (m, 1 H, ArH), 7.38–7.61 (m, 5 H, ArH), 7.98 (br. s, 1 H, NH) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta=35.2$, 38.5, 55.7, 55.8, 110.9, 111.1, 112.8, 119.5, 119.7, 119.9, 121.6, 124.8, 126.6, 127.4, 127.8, 128.9, 129.2, 132.6, 133.2, 135.8, 137.6, 138.1, 147.5, 148.8 ppm. MS: m/z (%) = 381 (40) [M†]. C₂₆H₂₃NO₂ (381.47): calcd. C 81.86, H 6.08, N 3.67%; found C 81.97, H 5.83, N 3.42%.

Diethyl $(1S^*,3aR^*,8bS^*)$ -1-(4-methoxyphenyl)-8b-methyl-1,3a,4,8btetrahydrocyclopenta[b]indole-3,3(2H)-dicarboxylate (11e): Yield 89% (0.75 g); colorless viscous liquid; R_f 0.35 (9:1 hexanes/EtOAc). IR (neat): $\tilde{v} = 3468$, 2363, 1726, 1597, 1516, 1351, 1263, 669 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 1.19$ (t, J = 7.4 Hz, 3 H, CH₃), 1.23 (t, J = 7.4 Hz, 3 H, CH₃), 1.47 (s, 3 H, CH₃), 2.22 (dd, J =13.6, 5.1 Hz, 1 H, $H^{2'}$), 2.57 (dd, J = 14.5, 13.1 Hz, 1 H, $H^{2''}$), 3.03 $(dd, J = 14.5, 5.1 \text{ Hz}, 1 \text{ H}, H^1), 3.70 \text{ (s, 3 H, OCH₃)}, 3.91 \text{ (br. s, 1)}$ H, NH), 4.06-4.28 (m, 4 H, $2 \times \text{CH}_2$), 4.64 (s, 1 H, H^{3a}), 5.53 (d, J = 7.5 Hz, 1 H, ArH), 6.24 (t, J = 7.4 Hz, 1 H, ArH), 6.41 (d, J =7.5 Hz, 1 H, ArH), 6.69 (d, J = 8.5 Hz, 2 H, ArH), 6.81 (d, J =8.5 Hz, 2 H, ArH), 6.82 (t, J = 7.8 Hz, 1 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 14.0, 14.1, 27.8, 36.3, 53.3, 55.1, 57.5, 61.2, 61.5, 65.8, 73.3, 108.6, 112.9, 117.7, 126.0, 127.4, 129.8, 130.4, 132.0, 150.4, 158.4, 169.1, 170.9 ppm. MS: m/z (%) = 423 (100) [M⁺]. C₂₅H₂₉NO₅ (423.50): calcd. C 70.90, H 6.90, N 3.31%; found C 70.73, H 7.13, N 3.11%.

 $(1S^*,3S^*,4aR^*,9aR^*)-3-[4-Methoxyphenvl]-1,2,3,4,9-pentahydro-$ 4a,9a-propanocarbazole Derivative 16: Yield 58% (0.49 g); viscous liquid; R_f 0.75 (6:1 hexanes/EtOAc). IR (neat): $\tilde{v} = 3446$, 2921, 2851, 2365, 1597, 1352, 1251, 1032, 669 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 1.11-1.25$ (m, 2 H, CH₂), 1.33-1.37 (m, 2 H, CH₂), 1.60-1.65 (m, 1 H, CH), 1.70-1.76 (m, 1 H, CH), 1.81-1.90 (m, 1 H, CH), 2.08-2.13 (m, 2 H, cyclohexane CH and H2'), 2.36 (q, J = 12.6 Hz, 1 H, $H^{2''}$), 3.14 (dd, J = 12.6, 5.9 Hz, 1 H, H^1), 3.67 (s, 3 H, OCH₃), 3.86 (dd, J = 12.6, 5.2 Hz, 1 H, H³), 4.23 (br. s, 1 H, NH), 5.72 (d, J = 7.5 Hz, 1 H, ArH), 6.17 (t, J = 7.4 Hz, 1 H, ArH), 6.42 (d, J = 7.5 Hz, 1 H, ArH), 6.59 (d, J = 8.5 Hz, 2 H, ArH), 6.73 (d, J = 8.5 Hz, 2 H, ArH), 6.80 (t, J = 7.5 Hz, 1 H, ArH), 7.41 (t, J = 7.4 Hz, 2 H, ArH), 7.50 (t, J = 7.0 Hz, 1 H, ArH), 7.91 (d, J = 8.3 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 17.8, 18.9, 32.7, 33.4, 34.8, 54.1, 55.1, 57.5, 59.4, 74.0, 107.6, 112.8, 116.8, 125.2, 127.3, 128.2, 128.6, 129.6, 130.0, 132.1, 133.0, 138.3, 151.2, 158.1, 201.7 ppm. MS: m/z (%) = 423 (100) [M⁺]. C₂₉H₂₉NO₂ (423.55): calcd. C 82.24, H 6.90, N 3.31 %; found C 82.53, H 6.71, N 3.51%.

(1*S**,3*S**,3*aR**,8*bR**)-1-Phenyl-1,2,3,4-tetrahydro-3a,8b-propanocy-clopenta[*b*]indole Derivative 19: Yield 25% (0.18 g); colorless crystals; m.p. 180–183 °C; R_f 0.61 (19:1 hexanes/EtOAc). IR (KBr): \tilde{v} = 3355, 3058, 2941, 2363, 1662, 1598, 1484, 1457, 1401, 1319, 1254, 1207, 1159, 1015, 746, 707 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 1.08–1.26 (m, 3 H, CH₂, CH), 1.50–1.56 (m, 1 H, CH), 1.62–1.69 (m, 2 H, CH₂), 1.91 (td, J = 13.0, 5.0 Hz, 1 H, H²'), 2.64 (q, J = 13.0 Hz, 1 H, H²''), 3.32 (dd, J = 13.0, 5.0 Hz, 1 H, H¹), 3.97 (dd,

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J = 13.0, 5.0 Hz, 1 H, H³), 6.54 (d, J = 7.8 Hz, 1 H, ArH), 6.67 (t, J = 7.3 Hz, 1 H, ArH), 6.94 (d, J = 7.5 Hz, 1 H, ArH), 7.00 (dt, J = 7.1, 1.0 Hz, 1 H, ArH), 7.23 (t, J = 7.1 Hz, 1 H, ArH), 7.31 (t, J = 7.4 Hz, 2 H, ArH), 7.39 (d, J = 7.3 Hz, 2 H, ArH), 7.46 (t, J = 7.5 Hz, 2 H, ArH), 7.54 (t, J = 7.4 Hz, 1 H, ArH), 8.11 (d, J = 7.8 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 25.1, 31.6, 36.8, 38.5, 52.4, 59.0, 69.1, 82.6, 108.9, 119.0, 124.0, 126.7, 127.8, 128.0, 128.3, 128.6, 128.91, 128.95, 133.1, 135.9, 137.8, 139.9, 201.3 ppm. MS: m/z (%) = 379 (100) [M†]. C₂₇H₂₅NO (379.49): calcd. C 85.45, H 6.64, N 3.69%; found C 85.21, H 6.91, N 3.92%.

 $(1S^*,3S^*,5aR^*,10bR^*)-1-(4-Methoxyphenyl)-1,2,3,4,5,6-hexahydro-$ 5a,10b-propanocyclohepta b indole Derivative 21: Yield 27% (0.23 g); colorless solid; m.p. 142-144 °C; R_f 0.67 (9:1 hexanes/ EtOAc). IR (KBr): $\tilde{v} = 3413$, 2930, 2854, 2363, 1676, 1604, 1512, 1486, 1448, 1363, 1248, 1222, 1179, 1035, 831, 742, 703 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 0.98–1.08 (m, 2 H, CH₂), 1.10–1.31 (m, 3 H, CH₂, CH), 1.40-1.55 (m, 2 H, CH₂), 1.78 (t, J = 14.4 Hz, 1 H, CH), 1.90–1.96 (m, 2 H, $H^{2'}$ and CH), 2.11 (dd, J = 14.4, 7.1 Hz, 1 H, CH), 2.30 (q, J = 12.9 Hz, 1 H, $H^{2''}$), 3.03 (dd, J =12.9, 5.1 Hz, 1 H, H¹), 3.66 (s, 3 H, OCH₃), 3.77 (dd, J = 12.9, 4.6 Hz, 1 H, H³), 4.65 (br. s, 1 H, NH), 5.45 (d, J = 7.3 Hz, 1 H, ArH), 6.14 (t, J = 7.3 Hz, 1 H, ArH), 6.47 (d, J = 7.6 Hz, 1 H, ArH), 6.61 (br. s, 4 H, ArH), 6.84 (t, J = 7.5 Hz, 1 H, ArH), 7.37 (t, J = 7.4 Hz, 2 H, ArH), 7.47 (t, J = 8.3 Hz, 1 H, ArH), 7.87 (d, J = 7.4 Hz, 2 H, ArH)J = 7.3 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta =$ 23.4, 25.1, 29.6, 31.2, 35.1, 36.4, 40.2, 55.1, 56.1, 58.7, 64.7, 106.6, 112.6, 115.6, 126.4, 127.3, 128.2, 128.5, 129.0, 130.4, 131.0, 133.0, 138.3, 151.0, 158.2, 202.2 ppm. MS: m/z (%) = 437 (100) [M⁺]. C₃₀H₃₁NO₂ (437.57): calcd. C 82.35, H 7.14, N 3.20%; found C 82.12, H 7.32, N 3.47%.

General Procedure for the [3+2] Cycloaddition of 2-(3,4-Dimethoxyphenyl)cyclopropyl Phenyl Ketone (2b) to Indoles 1a, 1c, and 1f in Presence of TiCl₄ as the Lewis Acid: A dilute solution of TiCl₄ (0.21 mL, 1.94 mmol) in dry CH₂Cl₂ (5 mL) was added dropwise at -78 °C to a stirred solution of 2b (0.50 g, 1.77 mmol) and appropriate indoles 1a, 1c or 1f (1.77 mmol) in CH₂Cl₂ (10 mL). After the complete addition of TiCl₄, the reaction mixture was stirred at -78 °C for 1-2 h. It was then diluted with ice-cold saturated NaHCO₃ solution (10 mL) and the white precipitate was filtered, washed with CHCl₃ (2×25 mL). The filtrate was washed with water (1×25 mL), followed by brine (25 mL) and dried with anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure to afford cyclopentannulated indoles 5a, 5c, and 5f, which were purified by column chromatography using silica gel and hexane/ EtOAc as eluent.

 $(1R^*,3S^*,3aR^*,8bS^*)$ -[1-(3,4-Dimethoxyphenyl)-1,2,3,3a,4,8b-hexahydrocyclopenta[b]indol-3-yl]phenylmethanone (5a): Yield 84% (0.59 g); colorless crystals; m.p. 145-146 °C; R_f 0.58 (3:1 hexanes/ EtOAc). IR (KBr): $\tilde{v} = 3337$, 1659, 1597, 1515, 1460, 1328, 1249, 1144, 1028, 756 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 2.21 (q, J = 12.2 Hz, 1 H, $H^{2''}$), 2.31 (td, J = 12.2, 6.1 Hz, 1 H, $H^{2'}$), 3.24 (m, 1 H, H¹), 3.85 (t, J = 10.1 Hz, 1 H, H^{8b}), 3.89 (s, 6 H, $2 \times OCH_3$, 4.02 (td, J = 12.2, 6.1 Hz, 1 H, H³), 4.80 (dd, J = 10.1, 6.1 Hz, 1 H, H^{3a}), 6.63–6.68 (m, 2 H, ArH), 6.84–6.87 (m, 4 H, ArH), 7.06 (t, J = 7.56 Hz, 1 H, ArH), 7.51 (t, J = 7.3 Hz, 2 H, ArH), 7.60 (t, J = 7.3 Hz, 1 H, ArH), 8.06 (dd, J = 8.3, 1.2 Hz, 2 H, ArH) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 39.8, 53.9, 55.6, 55.9, 56.0, 56.9, 66.8, 109.8, 110.5, 111.3, 118.8, 119.6, 124.3, 127.9, 128.6, 128.8, 131.6, 133.4, 134.9, 137.0, 147.8, 149.1, 149.7, 201.5 ppm. MS: m/z (%) = 399 (100) [M⁺]. $C_{26}H_{25}NO_3$ (399.48): calcd. C 78.17, H 6.31, N 3.51%; found C 78.30, H 6.12, N 3.69%.

Supporting Information (see also the footnote on the first page of this article): ¹H and ¹³C NMR spectroscopic data for all new compounds **3b**, **3c**, **5c**, **5f**, **4c**, **6e**′, **7e**′, **7e**′′, **9e**′, **11a**′, **11a**′′, **11c**′′, **12e**, **14**, **18**, **and 22**, and X-ray crystal structure data for compounds **3a** and **3d** are available.

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- [1] Review: P. S. Steyn, R. Vleggaar, Fortschr. Chem. Org. Naturst. 1985, 48, 1.
- [2] a) A. B. Smith III, R. Mewshaw, J. Am. Chem. Soc. 1985, 107, 1769; b) R. E. Mewshaw, M. D. Taylor, A. B. Smith III, J. Org. Chem. 1989, 54, 3449; c) A. B. Smith III, T. L. Leenay, J. Am. Chem. Soc. 1989, 111, 5761.
- [3] Y.-C. Kong, K.-F. Cheng, R. C. Cambie, P. G. Waterman, J. Chem. Soc., Chem. Commun. 1985, 47.
- [4] Y. C. Kong, K.-H. Ng, K.-H. Wat, A. Wong, I. F. Saxena, K.-F. Cheng, P. P. But, H.-T. Chang, Planta Med. 1985, 44, 304.
- [5] a) D. C. C. Wong, W. P. Fong, S. S. Lee, Y. C. Kong, K. F. Cheng, G. Stone, Eur. J. Pharmacol. 1998, 362, 87; b) K. F. Cheng, T. T. Wong, K. P. Chan, Y. C. Kong, Eur. J. Med. Chem. 1992, 27, 121.
- [6] T. W. Leung, G. Cheng, C. H. Chui, S. K. Ho, F. Y. Lau, J. K. Tjong, T. C. Poon, J. C. Tang, W. C. Tse, K. F. Cheng, K. C. Kong, *Chemotherapy* 2000, 46, 62.
- [7] K. R. Campos, M. Journet, S. Lee, E. J. J. Grabowski, R. D. Tillyer, J. Org. Chem. 2005, 70, 268.
- [8] H. Achenbach, K. Biemann, J. Am. Chem. Soc. 1965, 87, 4944.
- [9] a) P. Magnus, T. Gallagher, P. Brown, C. Huffman, J. Am. Chem. Soc. 1984, 106, 2105; b) T. Gallagher, P. Magnus, J. Am. Chem. Soc. 1983, 105, 2086.
- [10] M. E. Kuehne, P. J. Seaton, J. Org. Chem. 1985, 50, 4790.
- [11] a) M. Ishikura, K. Imaizumi, N. Katagiri, Heterocycles 2000, 53, 2201; b) J. Sapi, G. Massiot, The Chemistry of Heterocyclic Compounds: Bisindole Alkaloids, Supplement to vol. 25, part 4 (Ed.: J. E. Saxton), John Wiley & Sons, Chichester, 1994, pp. 625–623.
- [12] a) J. H. Sheu, C. A. Chen, B. H. Chen, Chem. Commun. 1999, 203; b) K. J. Henry, Jr., P. A. Grieco, J. Chem. Soc., Chem. Commun. 1993, 510; c) J.-H. Sheu, Y.-K. Chen, H.-F. Chung, P.-J. Sung, S.-F. Lin, Heterocycles 1996, 43, 1751 and references cited therein; d) K.-F. Cheng, K.-P. Chan, Y. C. Kong, D.-D. Ho, J. Chem. Soc., Perkin Trans. 1 1991, 2955; e) M. Ishikura, Heterocycles 1995, 41, 1385.
- [13] W.-L. Chan, D.-D. Ho, C. P. Lau, K.-H. Wat, Y.-C. Kong, K.-F. Cheng, T.-T. Wong, K.-P. Chan, Eur. J. Med. Chem. 1991, 26, 387.
- [14] O. Miyata, Y. Kimura, T. Naito, Chem. Commun. 1999, 2429.
- [15] A. Ganesan, C. H. Heathcock, Tetrahedron Lett. 1993, 34, 439.
- [16] S. J. Martinez, I. Dalton, J. A. Joule, *Tetrahedron* 1984, 40, 3339.
- [17] a) J. Bergman, L. Venemalm, *Tetrahedron Lett.* 1987, 28, 3471;
 b) J. Bergman, L. Venemalm, *Tetrahedron* 1992, 48, 759;
 c) J. Bergman, L. Venemalm, A. Gogoll, *Tetrahedron* 1990, 46, 6067;
 d) J. Bergman, L. Venemalm, *Tetrahedron Lett.* 1988, 29, 2993;
 e) J. Bergman, L. Venemalm, *Pure Appl. Chem.* 1994, 66, 2331.
- [18] M. Ishikura, K. Imaizumi, N. Katagiri, Heterocycles 2000, 53, 553.
- [19] C. A. Harrison, R. Leineweber, C. J. Moody, J. M. J. Williams, Tetrahedron Lett. 1993, 34, 8527.
- [20] M. A. Kerr, R. G. Keddy, Tetrahedron Lett. 1999, 40, 5671.
- [21] D. B. England, T. D. O. Kuss, R. G. Keddy, M. A. Kerr, J. Org. Chem. 2001, 66, 4704.
- [22] P. Harrington, M. A. Kerr, Tetrahedron Lett. 1997, 38, 5949.
- [23] See also: D. B. England, T. K. Woo, M. K. Kerr, Canad. J. Chem. 2002, 80, 992.

FULL PAPER

- [24] a) B. Patro, B. Deb, H. Ila, H. Junjappa, J. Org. Chem. 1992, 57, 2257; b) B. Patro, B. Deb, H. Ila, H. Junjappa, Tetrahedron 1994, 50, 255; c) P. K. Patra, B. Patro, H. Ila, H. Junjappa, Tetrahedron Lett. 1993, 34, 3951; d) P. K. Patra, V. Sriram, H. Ila, H. Junjappa, Tetrahedron 1998, 54, 531; e) U. K. Syam Kumar, P. K. Patra, H. Ila, H. Junjappa, J. Chem. Soc., Perkin Trans. 1 2000, 1547; f) P. K. Mohanta, S. Peruncheralathan, H.
- Ila, H. Junjappa, *J. Org. Chem.* **2001**, *66*, 1503; g) S. Peruncheralathan, V. Sriram, H. Ila, H. Junjappa, *Tetrahedron* **2004**, *60*, 5603
- [25] E. J. Corey, M. Chaykovsky, J. Am. Chem. Soc. 1965, 87, 1353.
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